

43.) $V_{\ddot{o}} = 44,8 \text{ dm}^3$

$P = 200 \text{ W}$

$P_1 = 10^5 \text{ Pa}$

egyatomos gáz $\rightarrow f = 3$

$V_{b1} = V_{j1} = \frac{V_{\ddot{o}}}{2}$

$V_{j2} = \frac{V_{j1}}{2} \rightarrow t = ?$

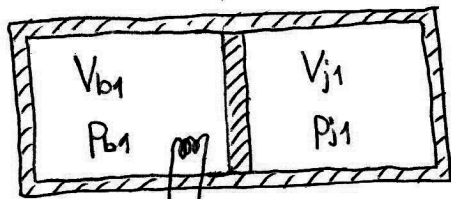
$E_b = \frac{f}{2} nRT = \frac{f}{2} pV$

$\Delta E_b = Q + W$

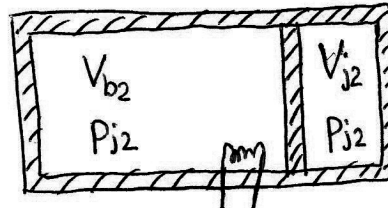
$Q = P \cdot t$

$k = \frac{f+2}{f}$

$PV^k = \text{all}$



$t = ?$



$P_1 = P_{b1} = P_{j1}$

nyomás
meg egyezik

$P_{j2} = P_{b2} = P_2$

A teljes belső energia változás:

$\Delta E_{b\ddot{o}} = Q + W_{\ddot{o}}$

Q csak a fűtőszál miatt!

$\Delta E_{bb} + \Delta E_{bj} = Q + \underbrace{W_b + W_j}_0$

$W_b = -W_j$ egymáson végzik!

$\frac{3}{2}(P_{b2}V_{b2} - P_{b1}V_{b1}) + \frac{3}{2}(P_{j2}V_{j2} - P_{j1}V_{j1}) = Q = P \cdot t$

$\frac{3}{2}\left(P_2 \frac{3}{4}V_{\ddot{o}} - P_1 \frac{V_{\ddot{o}}}{2} + P_2 \frac{V_{\ddot{o}}}{4} - P_1 \frac{V_{\ddot{o}}}{2}\right) = P \cdot t$

$\frac{3}{2}(P_2V_{\ddot{o}} - P_1V_{\ddot{o}}) = P \cdot t$

$t = \frac{\frac{3}{2}(P_2V_{\ddot{o}} - P_1V_{\ddot{o}})}{P} = \dots = \underline{\underline{73,1 \text{ s}}}$

adiabaticus

$P_1V_1^k = P_2V_2^k$ jobb
oldalra

$P_1V_{j1}^k = P_2V_{j2}^k$

$P_2 = P_1 \left(\frac{V_{j1}}{V_{j2}}\right)^k$